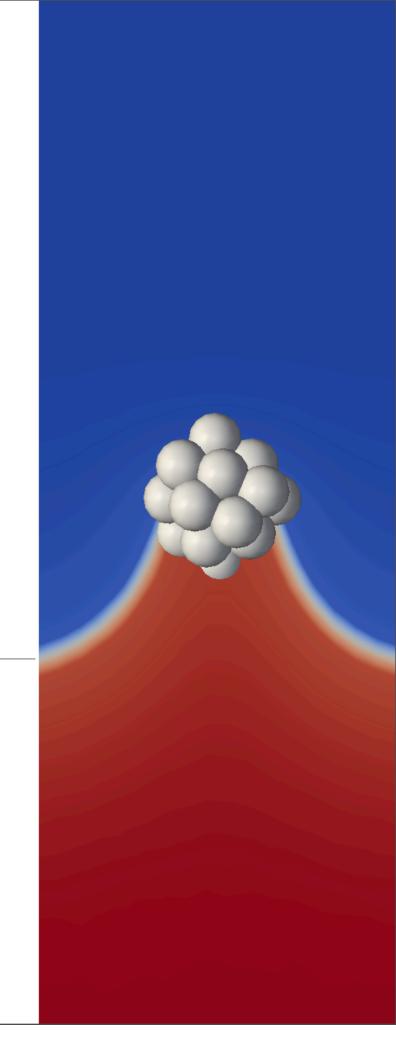


Modeling two-phase flow

Marcello Sega

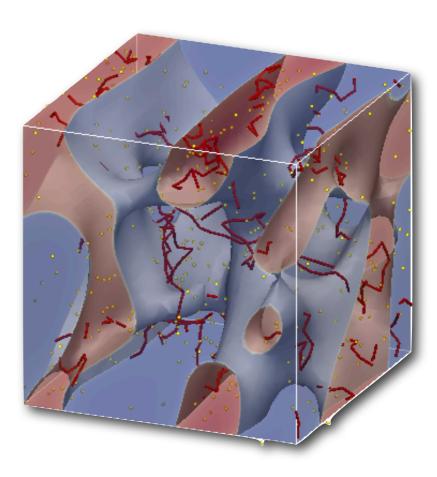
ESPResSo Summer School 2013, Stuttgart

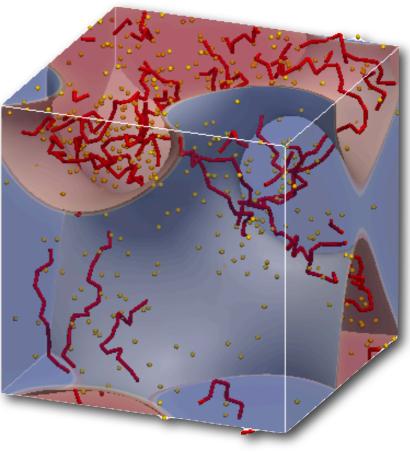


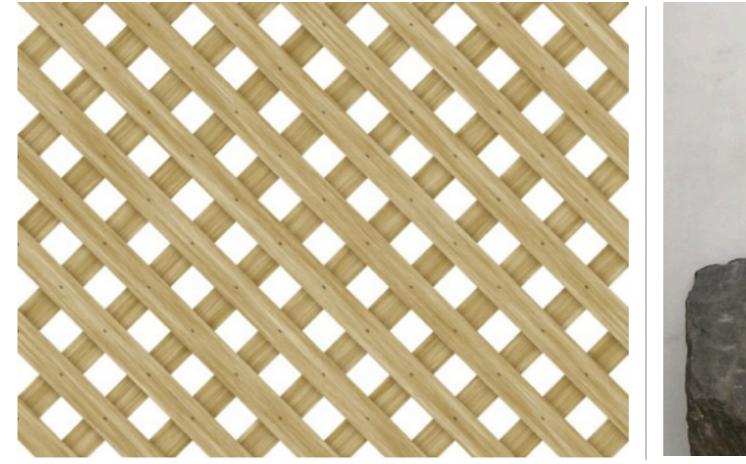
Tuesday, October 8, 2013

What you will see

- Short intro to Lattice-Boltzmann
- Shan-Chen multicomponent fluid
- ESPResSo implementation
- Coupling particle dynamics
- Examples







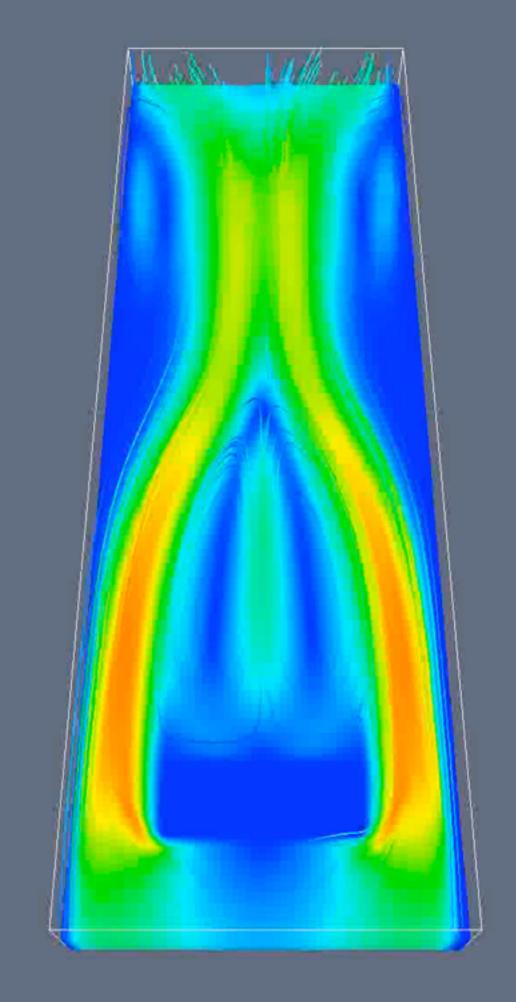


The Lattice-Boltzmann Method

Tuesday, October 8, 2013

What is LB and why would I use it?

- •LB = Boltzmann transport equation on a lattice
- Solves Navier-Stokes equations over a wide range of Reynolds numbers
- •Simple parallel code
- •Coarse-graining solvent degrees of freedom (more on this later)



Video: ESPResSo LB-GPU, D. Roehm, ICP

From the Boltzmann Equation to Hydrodynamics

$$\frac{\partial}{\partial t}f + \mathbf{v} \cdot \frac{\partial}{\partial \mathbf{r}}f = -\lambda\left(f - f^{\text{eq}}\right) \qquad f^{\text{eq}}(\mathbf{v}) = \left(\frac{m}{2\pi k_B T}\right)^{\frac{3}{2}} \frac{\rho}{m} \exp\left[-\frac{m(\mathbf{v} - \mathbf{u})^2}{2k_B T}\right]$$

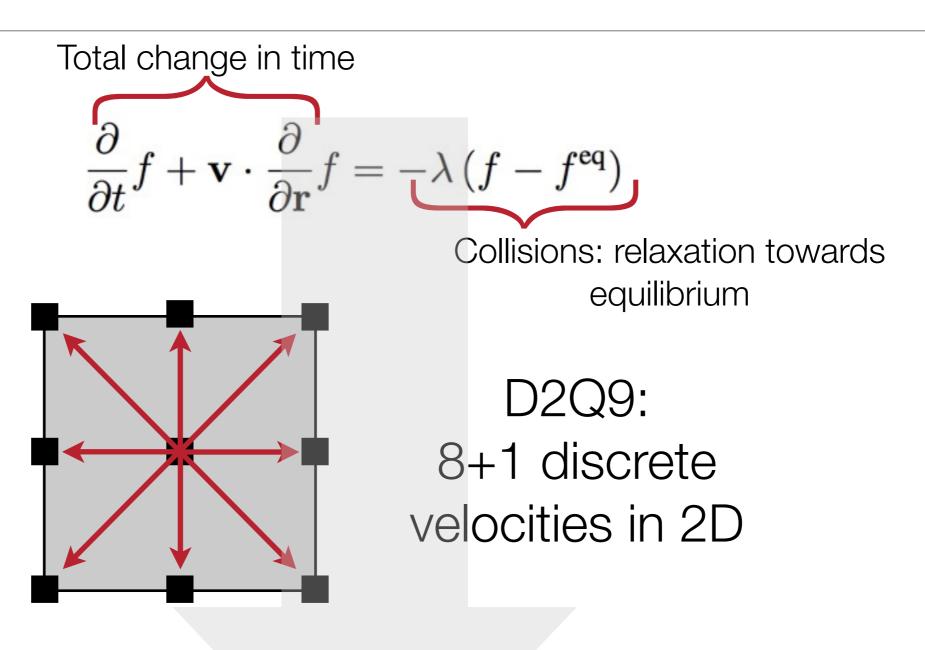
Momenta of the distribution f

$$m \int f d\mathbf{v} = \rho(\mathbf{r}, t)$$
 $m \int \mathbf{v} f d\mathbf{v} = \rho \mathbf{u}(\mathbf{r}, t)$

$$m\int \frac{\mathbf{v}^2}{2}f\,d\mathbf{v} = \rho e(\mathbf{r},t)$$

$$\begin{aligned} \frac{\partial}{\partial t}\rho + \frac{\partial}{\partial \mathbf{r}}\cdot(\rho\mathbf{u}) &= 0, \\ \frac{\partial}{\partial t}(\rho\mathbf{u}) + \frac{\partial}{\partial \mathbf{r}}\cdot(\rho\mathbf{u}\otimes\mathbf{u}) &= -\frac{\partial}{\partial \mathbf{r}}p + \frac{\partial}{\partial \mathbf{r}}\cdot\sigma + \mathbf{g}. \end{aligned}$$

Easy LB: the BGK approximation

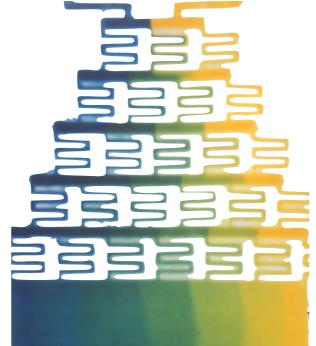


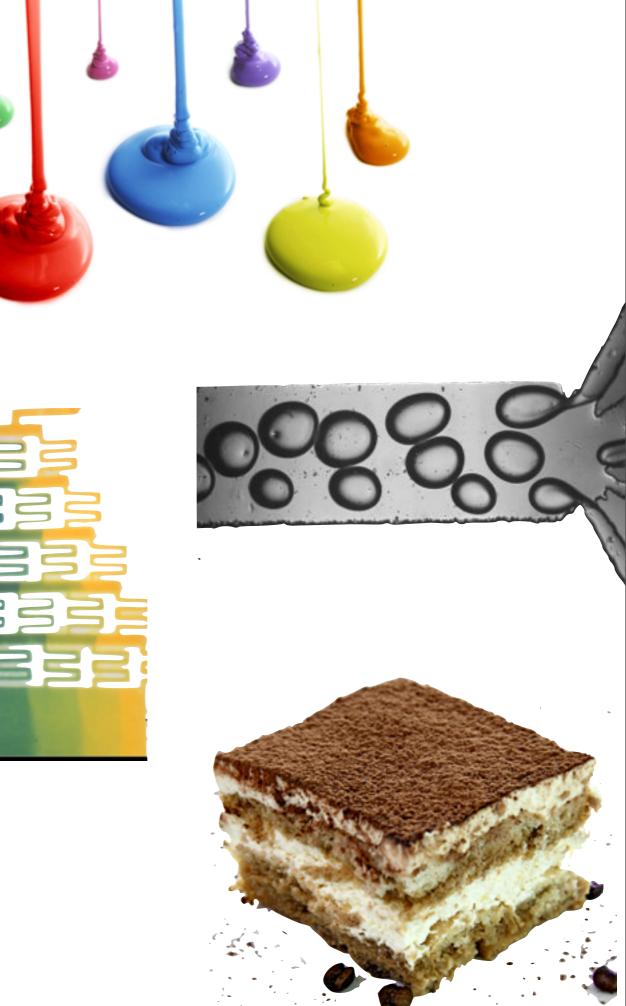
 $f_i(\mathbf{r} + \tau \mathbf{c}_i, t + \tau) = f_i(\mathbf{r}, t) - \lambda \left[f_i(\mathbf{r}, t) - f_i^{eq}(\mathbf{r}, t) \right]$

Discretized Boltzmann Equation

Multicomponent Fluids

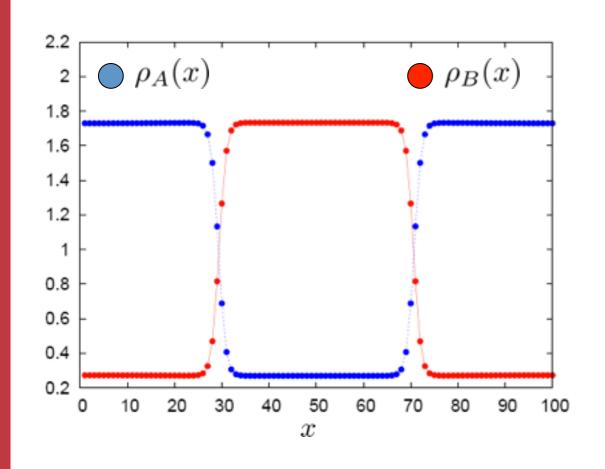
- Emulsification
- Encapsulation
- Sprays
- Food processing
- Paints
- Oil recovery



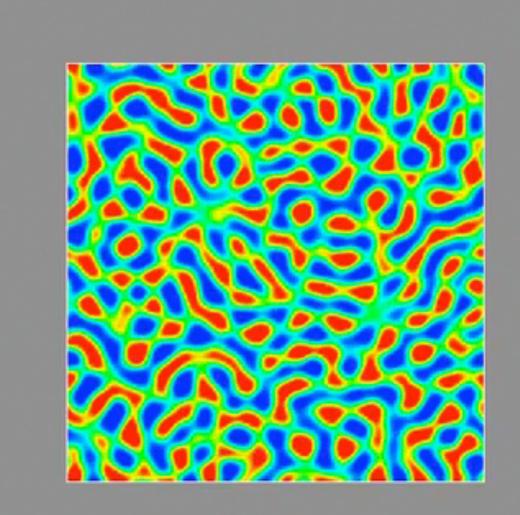


The Shan-Chen multicomponent fluid

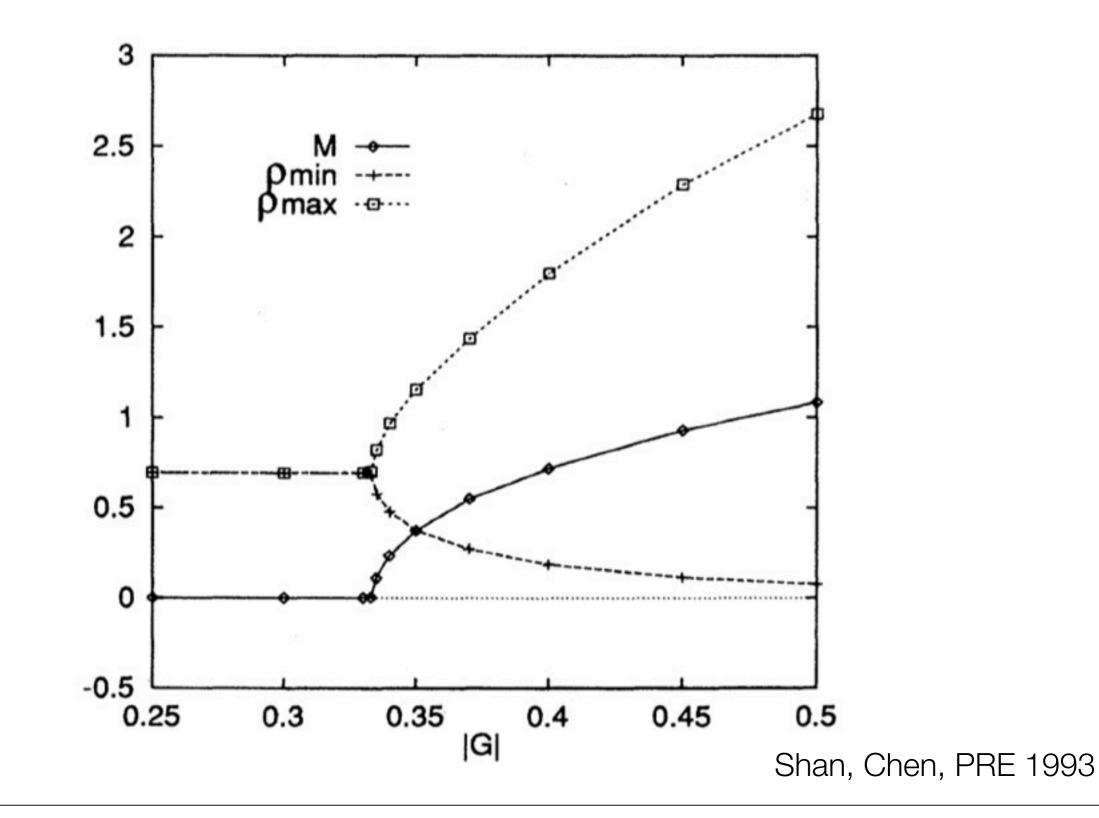
- Two density fields (index ζ)
- One baricentric velocity
- Force **g** between components



$$egin{aligned} m{g}_{\zeta}(m{r}) &= -
ho_{\zeta}(m{r}) \sum_{m{r}'} \sum_{\zeta'} g_{\zeta\zeta'}
ho_{\zeta'}(m{r}')(m{r}'-m{r}) \ m{g}_{\zeta}(m{r}) &\simeq -
ho_{\zeta}(m{r}) \sum_{\zeta'} g_{\zeta\zeta'}
abla
ho_{\zeta'}(m{r}). \end{aligned}$$



The multi-phase state diagram



The Shan-Chen multicomponent fluid

Continuity

$$\partial_t \rho + \boldsymbol{\nabla} \cdot (\rho \boldsymbol{u}) = 0$$

Navier-Stokes
$$\rho\left(\frac{\partial}{\partial t}\boldsymbol{u} + (\boldsymbol{u}\cdot\boldsymbol{\nabla})\boldsymbol{u}\right) = -\boldsymbol{\nabla}p + \boldsymbol{\nabla}\cdot(\boldsymbol{\Pi} + \hat{\boldsymbol{\sigma}}) + \sum_{\zeta}\boldsymbol{g}_{\zeta}$$

Components diffusion

$$rac{\partial}{\partial t}
ho_{\zeta} + oldsymbol{
abla} \cdot (
ho_{\zeta}oldsymbol{u}) = oldsymbol{
abla} \cdot (oldsymbol{D}_{\zeta} + \hat{oldsymbol{\xi}}_{\zeta})$$

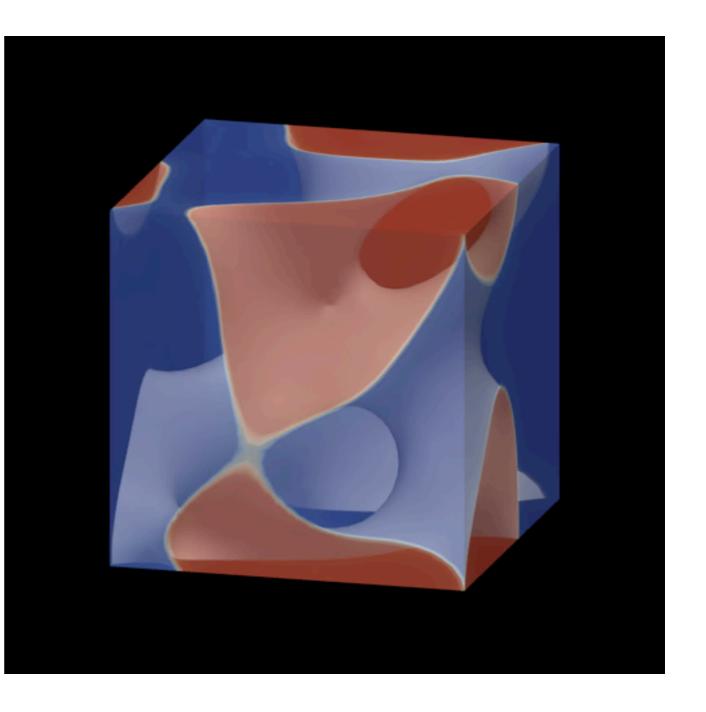
$$\rho = \sum_{\zeta} \rho_{\zeta} \qquad p = \sum_{\zeta} p_{\zeta} = \sum_{\zeta} c_s^2 \rho_{\zeta} \qquad u = \frac{1}{\rho} \sum_{\zeta} \sum_{i} f_{\zeta i} c_i + \frac{1}{2\rho} \tau g$$

Modes evolution and transport coefficients

19 modes: $m_{\zeta k}^* = (1 + \lambda_k)m_{\zeta k} + m_{\zeta k}^g + \phi_k r_k$ 1 mass, 3 momentum 5 shear + 1 bulk stress 9 ghost (kinetic) $\rho\left(\frac{\partial}{\partial t}\boldsymbol{u} + (\boldsymbol{u}\cdot\boldsymbol{\nabla})\boldsymbol{u}\right) = -\boldsymbol{\nabla}p + \boldsymbol{\nabla}\cdot(\boldsymbol{\Pi} + \hat{\boldsymbol{\sigma}}) + \sum_{\boldsymbol{v}}\boldsymbol{g}_{\boldsymbol{\zeta}}$ $\frac{\partial}{\partial t}\rho_{\zeta} + \boldsymbol{\nabla} \cdot (\rho_{\zeta}\boldsymbol{u}) = \boldsymbol{\nabla} \cdot (\boldsymbol{D}_{\zeta} + \hat{\boldsymbol{\xi}}_{\zeta})$ $\eta_s = -\rho c_s^2 \tau \left(\frac{1}{\lambda_s} + \frac{1}{2}\right) \qquad \mu = -\tau \left(\frac{1}{\lambda_M} + \frac{1}{2}\right)$ $\eta_b = ho c_s^2 au \left(rac{1}{\lambda_b}+rac{1}{2}
ight)$ $\Pi_{\alpha\beta} = \eta_s \bigg(\frac{\partial}{\partial r_\alpha} u_\beta + \frac{\partial}{\partial r_\beta} u_\alpha - \frac{2}{3} \frac{\partial}{\partial r_\gamma} u_\gamma \delta_{\alpha\beta} \bigg)$ $D_{\zeta\alpha} = \mu \left[\left(\frac{\partial p_{\zeta}}{\partial r} - \frac{\rho_{\zeta}}{\rho} \frac{\partial p}{\partial r} \right) - \left(g_{\zeta\alpha} - \frac{\rho_{\zeta}}{\rho} g_{\alpha} \right) \right]$ $+ \eta_b rac{\partial}{\partial r_{\sim}} u_\gamma \delta_{lphaeta},$

Shan-Chen in ESPResSo: some technical details

- GPU code (all examples done on this laptop)
- Full 3D multi-relaxation times (instead of simple BGK)
- Written in mode space (evolving modes rather than populations)
- Solves Fluctuating Hydrodynamics (all modes are thermalized)
- Extension of the Dünweg/Ladd/ Schiller Generalized Lattice Gas



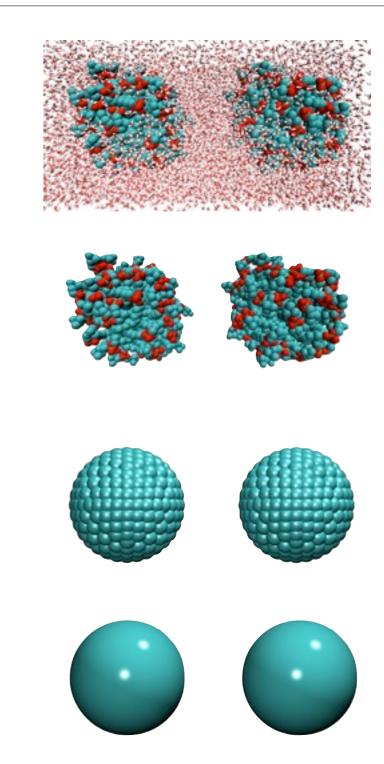
Lattice-Boltzmann as a thermostat: Solvent Coarse-Graining

Full Atomistic

Solvent Removal (e.g.+GB electrostatics)

Coarse-graining

"Extreme" coarse-graining

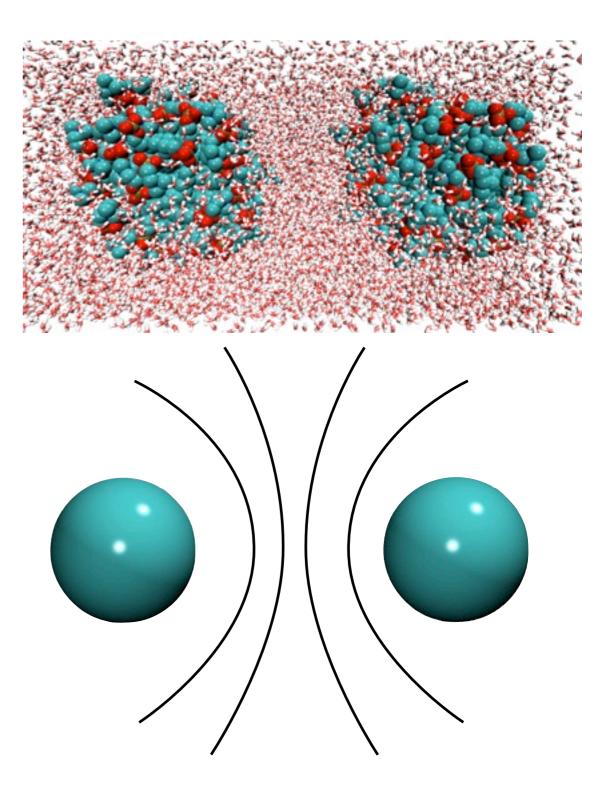


Lattice-Boltzmann as a thermostat: Solvent Coarse-Graining

- •Long range interactions might play a role
 - •Electrostatics
 - Hydrodynamic interaction
- •Usual thermostats are neither momentum-preserving, nor local
- •Example: Nosè-Hoover thermostat non Galileian-invariant: momentum preserved only when COM is at rest non local: momentum preserved only globally

$$m_i \mathbf{d} r_i / dt = \mathbf{p}_i, \quad \frac{d\mathbf{p}_i}{dt} = -\nabla_i U - \alpha \mathbf{p}_i$$

$$\frac{d\alpha}{dt} = \frac{1}{t_s}(T - T_0)$$



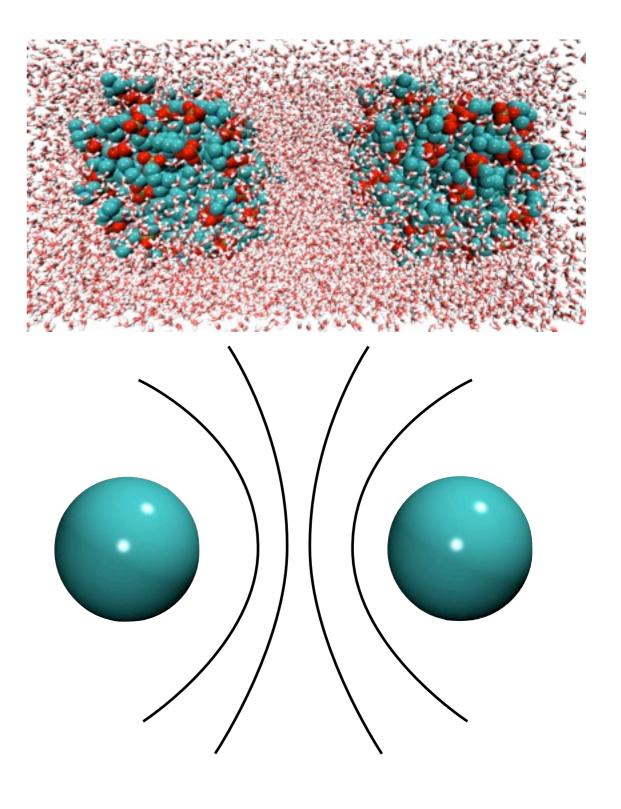
Lattice-Boltzmann as a thermostat: Solvent Coarse-Graining

•Several options: DPD, SRD, LB

•for LB: Ahlrichs-Dünweg coupling

$$m\boldsymbol{a}_i = \boldsymbol{F} - \gamma \left[\boldsymbol{v}_i - \boldsymbol{u}(\boldsymbol{r}_i) \right] + \boldsymbol{R}$$

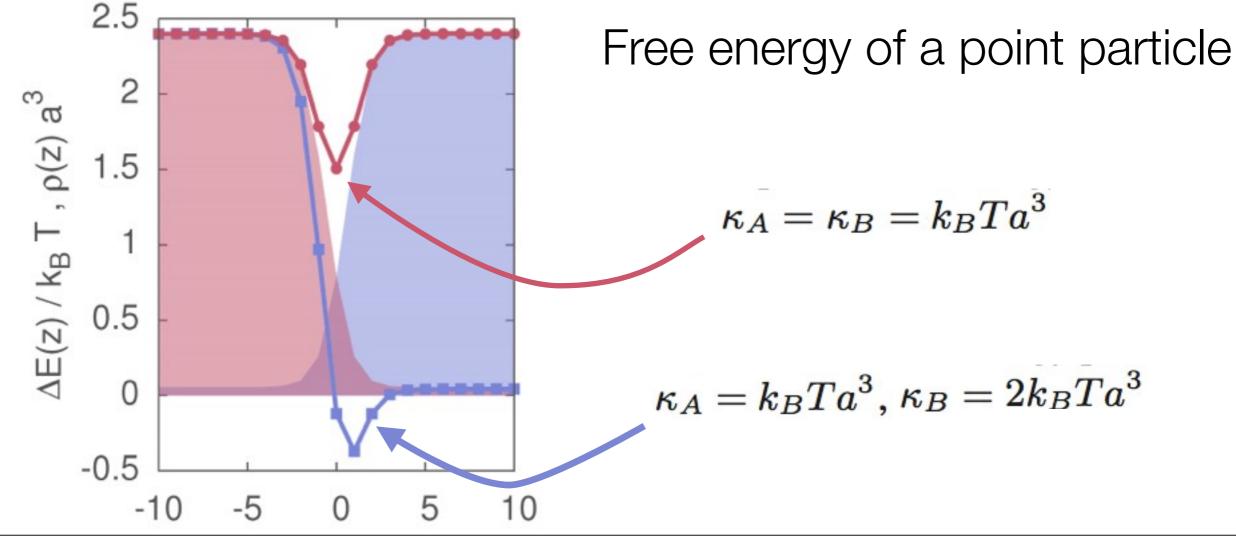
- •Langevin eq. guarantees proper thermalization
- Provides a meaningful coupling to the fluid



Shan-Chen LB as a thermostat: Solvent Coarse-Graining

- A simple extension is not enough
- Fluid-particle interaction: solvation free energy

$$egin{aligned} mm{a}_i &= m{F} - \gamma \left[m{v}_i - m{u}(m{r}_i)
ight] + m{R} \ m{F}_i^{ ext{ps}} &= -\sum_{\zeta} \kappa_{\zeta} m{
abla}
ho_{\zeta}(m{r}_i) \end{aligned}$$



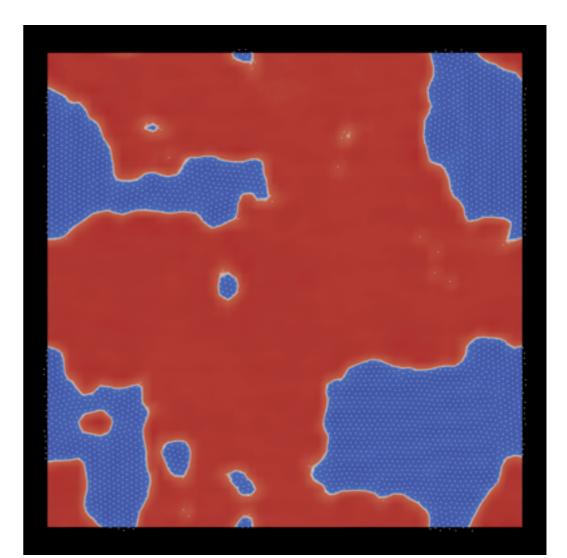
Shan-Chen LB as a thermostat: Solvent Coarse-Graining

- A problem of symmetry...
- Particle-fluid interaction

$$oldsymbol{F}_{\zeta}^{\mathrm{fs}}(oldsymbol{r}) = -\lambda_{\zeta}
ho_{\zeta}(oldsymbol{r})\sum_{i,oldsymbol{r}'}\Theta\left[rac{(oldsymbol{r}_i-oldsymbol{r})}{|oldsymbol{r}_i-oldsymbol{r}|}\cdotrac{(oldsymbol{r}'-oldsymbol{r})}{|oldsymbol{r}'-oldsymbol{r}|}
ight]rac{oldsymbol{r}'-oldsymbol{r}}{|oldsymbol{r}'-oldsymbol{r}|^2}$$

 $\Theta(x) = 1$ if 0 < x < 1 and 0 otherwise.

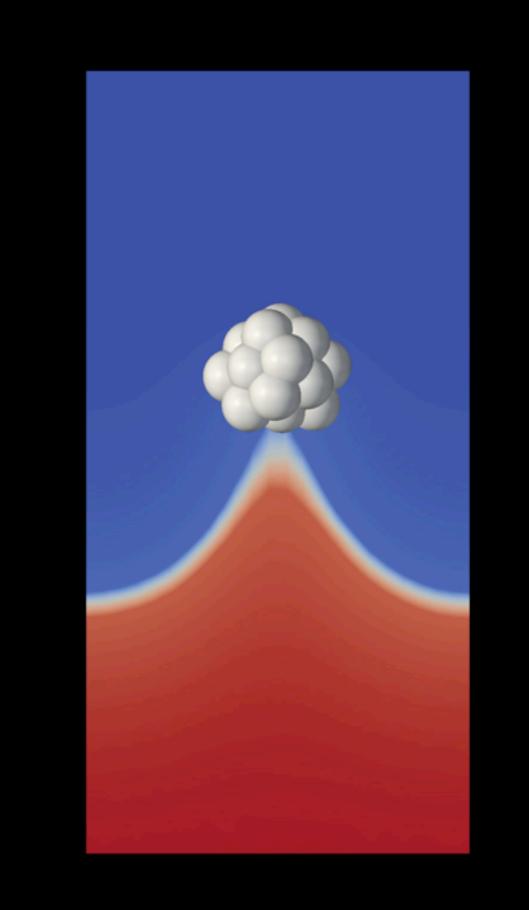
- For small λ s, same particle free energy
- Density change around the particle (excluded volume/hydration effect)



Only one fluid here!

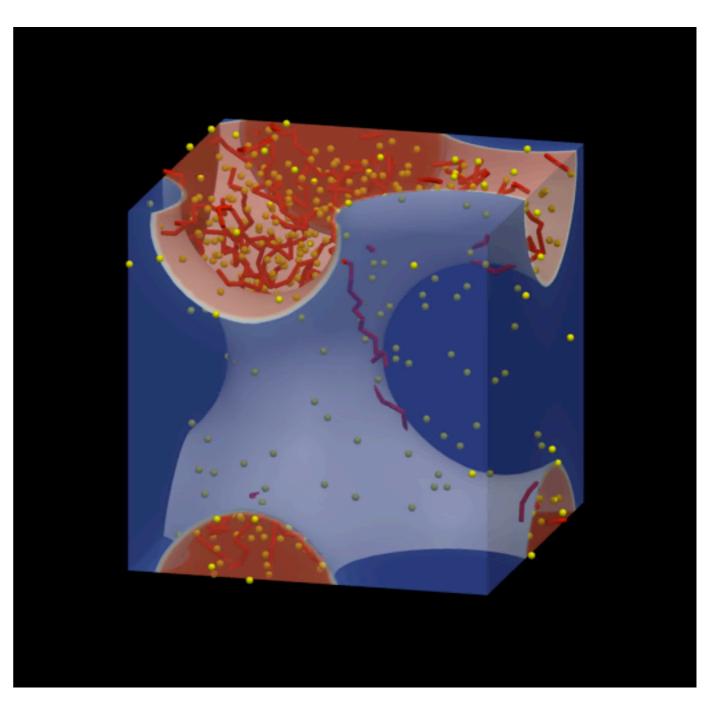
Effect of solvation

- Raspberry colloid
- $\lambda_{red} < 0$ ("red" component attracted)
- $\lambda_{\text{blue}} > 0$ ("blue" component repelled)
- Interface protrusion



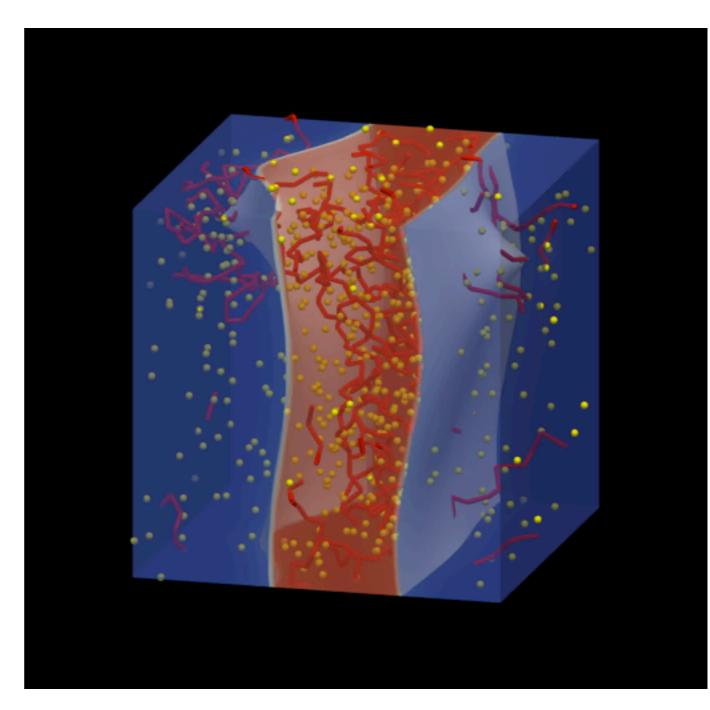
Electrostatic stabilization (I)

- Polyelectrolytes in a bicomponent fluid
- 10 chains (64 monomers each)
 + 640 counterions
- P3M electrostatics
- 32x32x32 SC grid
- κ_{red}<0, same κ_{red} for monomers and counterions



Electrostatic stabilization (II)

- Polyelectrolytes in a bicomponent fluid
- 10 chains (64 monomers each)
 + 640 counterions
- Debye-Hückel screened electrostatics
- 32x32x32 SC grid
- κ_{red}<0, same κ_{red} for monomers and counterions

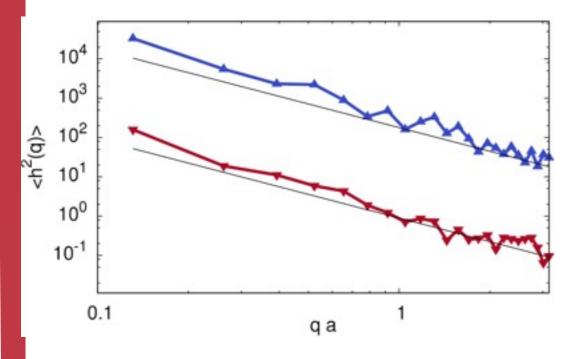


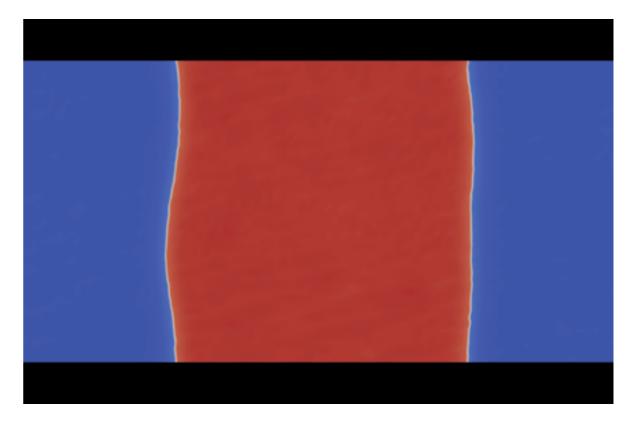
$$U_{ij}^{DH}(r_{ij}) = \begin{cases} q_i q_j \ell_B \exp(-\kappa r_{ij})/r_{ij} & r_{ij} < r_c \\ 0 & r_{ij} \ge r_c \end{cases}$$

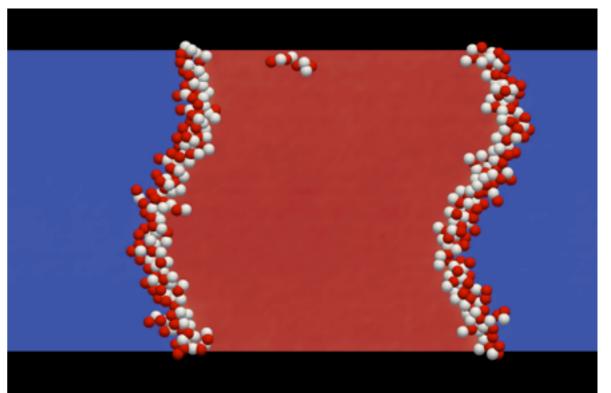
Surfactants & Surface Tension

- Can we achieve proper thermalization?
- Simulate amphiphilic dumbbells
- Measure interface fluctuation spectrum

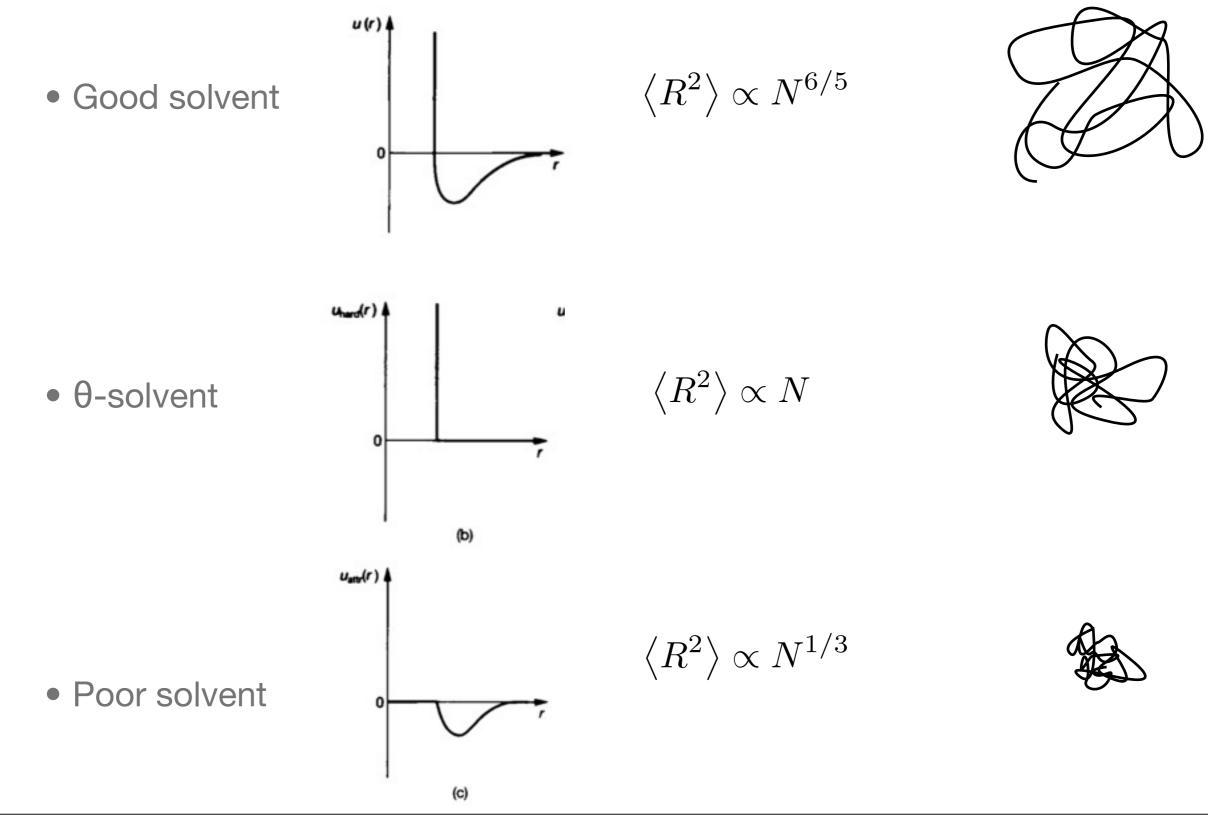
$$\left\langle h^2(q) \right\rangle = \frac{2k_BT}{\gamma_{AB}} \frac{1}{q^2}$$







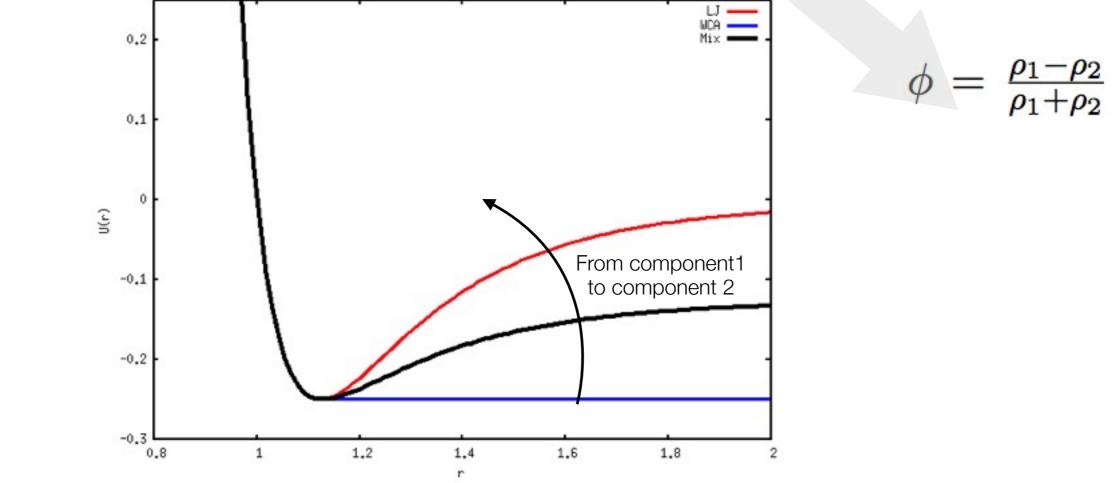
Solvent Affinity



Solvent affinity

d)

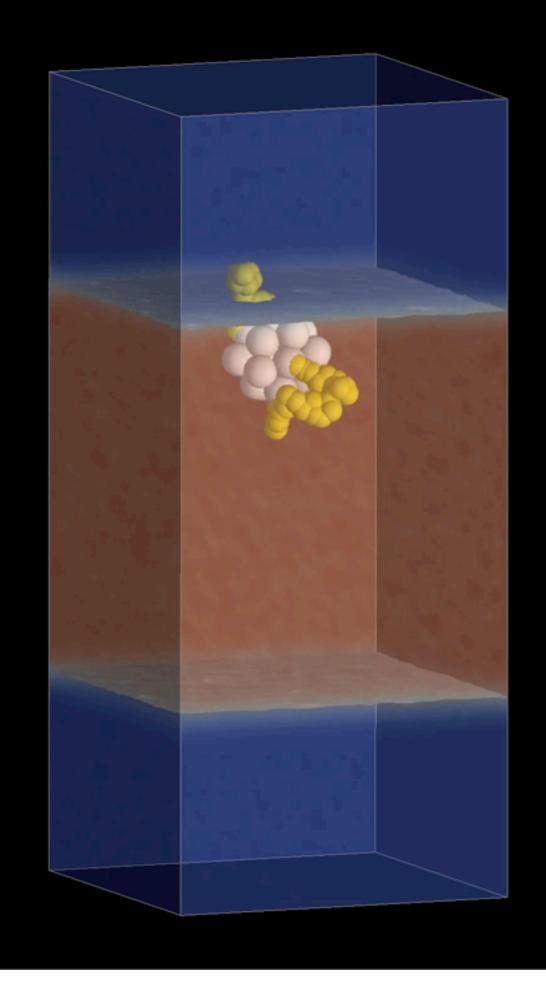
$$\begin{split} V_{\rm LJ}(r) &= \begin{cases} 4\epsilon ((\frac{\sigma}{r-r_{\rm off}})^{12} - (\frac{\sigma}{r-r_{\rm off}})^6 + c_{\rm shift}) &, \text{if } r_{\rm min} + r_{\rm off} < r < r_{\rm cut} + r_{\rm off} \\ 0 &, \text{otherwise} \end{cases} \\ X \\ A(r) &= \begin{cases} \frac{(1-\alpha_1)}{2} [1 + \tanh(2\phi)] + \frac{(1-\alpha_2)}{2} [1 + \tanh(-2\phi)] &, \text{if } r > r_{\rm cut} + 2^{\frac{1}{6}}\sigma \\ 1 &, \text{otherwise} \end{cases} \end{split}$$



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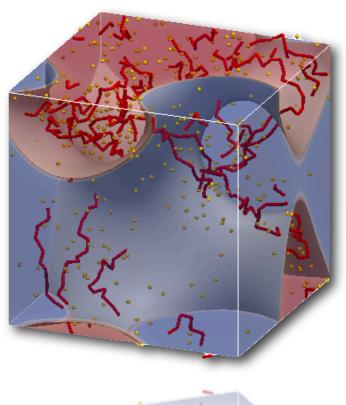
Solvent affinity

- Colloid with two polymer arms
- "Stretched" start
- Blue: bad solvent
- Red: good solvent



Conclusions

- Newly implemented Shan-Chen bicomponent fluid
- Multi-relaxation times, Fluctuating Hydrodynamics
- Coupling with particles:
 - Solvation free energy
 - Component-dependent forces
- Complex fluid-fluid interfaces where thermal energy competes with surface tension



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- S. Chen and G. Doolen, *Annu. Rev. Fluid Mech.*,1998, **30**,329–364
- Visualization software: VMD (http://www.ks.uiuc.edu/Research/vmd/) ParaView (http://www.paraview.org
- ESPResSo

